Hermitian Operators – Elementary Ideas, Quantum Mechanical Operator for Linear Momentum, Angular Momentum and Energy as Hermitian Operator

It is a quite well-known fact that all the physical properties are actually real quantities, and therefore are bound to have real values. It means that any operator which is used to represent a physical property must yield real values. In this section, we will discuss the elementary idea of Hermitian operators (named in honor of a great mathematician Charles Hermite), and will also prove that many important operators in quantum mechanics like linear momentum, angular momentum and Hamiltonian are Hermitian in nature.

> Elementary Idea of Hermitian Operator

Every physical property must have real eigen or expectation values, which therefore implies that the corresponding operators should have some special characteristics. One of the most important special characteristics includes a feature that the Hermitian conjugate of such an operator should be itself. In other words, if the Hermitian conjugate of an operator is itself, the operator is called as Hermitian; however, if the Hermitian conjugate of an operator is equal to its negative expression, the operator is called as anti-Hermitian or skew-Hermitian. Mathematically, we can say that

$$if A^{\dagger} = A; \qquad A \text{ is Hermitian} \tag{316}$$

$$if A^{\dagger} = -A; \qquad A \text{ is anti-Hermitian} \qquad (317)$$

Where A is an operator whose Hermitian conjugate is represented by A^{\dagger} .

However, the obvious question regarding the aforementioned definition would be "what is a Hermitian conjugate and how is it obtained". The answer is "the operator A^{\dagger} will be called as the Hermitian conjugate (or adjoint) of operator A if the operation of A^{\dagger} on the complex conjugate of function ψ gives the same result as when the A is operated over ψ ". Mathematically, we can say that

$$\langle \psi | A | \psi \rangle = \int_{-\infty}^{+\infty} \psi^*(x) A \psi(x) dx = \langle \psi | A \psi \rangle = \langle A^{\dagger} \psi | \psi \rangle$$
(318)

or

$$\left\langle A^{\dagger}\varphi|\psi\right\rangle = \left\langle \varphi|A\psi\right\rangle \tag{319}$$

1. Hermitian conjugates of different operators: The Hermitian conjugates of different operators can be studied in three different categories.

i) Hermitian conjugates of quantum mechanical operators:

Let Q be any quantum mechanical operator, then by the definition of Hermitian conjugates operator, we have the following condition.



$$\langle \varphi | Q \psi \rangle = \left\langle Q^{\dagger} \varphi | \psi \right\rangle \tag{320}$$

If Q is the momentum operator, then we can proceed as discussed below.

$$\int \psi^* \hat{p}_x \psi dx = \int \psi \hat{p}_x \psi^* dx \tag{321}$$

$$\int \psi^* \left(\frac{h}{2\pi i} \frac{\partial}{\partial x}\right) \psi dx = \int \psi \left(\frac{h}{2\pi i} \frac{\partial}{\partial x}\right)^{\dagger} \psi^* dx$$
(322)

$$\int \psi \left(\frac{h}{2\pi i}\frac{\partial}{\partial x}\right)^{\dagger} \psi^* dx = \int \psi \left(\frac{h}{2\pi i}\right)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger} \psi^* dx$$
(323)

$$\int \psi \left(\frac{h}{2\pi i} \frac{\partial}{\partial x}\right)^{\dagger} \psi^* dx = \int \psi \left(-\frac{h}{2\pi i}\right) \left(-\frac{\partial}{\partial x}\right) \psi^* dx$$
(324)

$$\int \psi \left(\frac{h}{2\pi i} \frac{\partial}{\partial x}\right)^{\dagger} \psi^* dx = \int \psi \left(\frac{h}{2\pi i} \frac{\partial}{\partial x}\right) \psi^* dx$$
(325)

Therefore, we can say that the Hermitian conjugate of the linear momentum operator is itself, and hence it is a Hermitian operator. Now from the most primitive definition of Hermitian operators, that all operators which correspond to observable quantities, we can say that the Hermitian conjugates of the following operator are themselves.

www.dalalinstitute.comHermitian conjugate Operator x x \hat{x}^2 \hat{x}^2 \hat{p}_x \hat{p}_x \hat{p}_{r}^{2} \hat{p}_{r}^{2} \hat{T}_x \hat{T}_{r} $\hat{V}(x)$ $\hat{V}(x)$ Ĥ Ĥ

ii) Hermitian conjugates of a constant operator:

There are some operators which are complex numbers. The Hermitian conjugates of such operators are actually their complex conjugates. Let we have the operator A

$$\hat{A} = a + ib \tag{326}$$



and since the definition of Hermitian operator is

$$\langle \varphi | A \psi \rangle = \langle A^{\dagger} \varphi | \psi \rangle \tag{327}$$

gives the integer as

$$\langle \varphi | (a+ib)\psi \rangle = \langle (a-ib)\varphi | \psi \rangle = (a+ib)\langle \varphi | \psi \rangle$$
(328)

Hence, the Hermitian conjugates of constant operators are their complex conjugates. The Hermitian conjugates of some operators are given below.



Integrating the above equation by part, we get

$$\left\langle \varphi \left| \frac{d}{dx} \psi \right\rangle = \left[\varphi^*(x) \psi(x) \right] - \int_{-\infty}^{+\infty} \frac{d\varphi^*(x)}{dx} \psi(x) \, dx \tag{328}$$

$$= 0 - \left\langle \frac{d}{dx} \varphi \middle| \psi \right\rangle \tag{329}$$

$$= -\left\langle \frac{d}{dx}\varphi \middle|\psi\right\rangle \tag{330}$$

Hence, the Hermitian conjugate of d/dx operator is -d/dx. Similarly, we can prove that the Hermitian conjugate of d^2/dx^2 is d^2/dx^2 .



2. Properties of Hermitian conjugates: From the definition and properties of scalar product, adjoints or Hermitian conjugate show the following properties.

i) Let *C* a constant and *A* as an operator.

$$(CA)^{\dagger} = C^* A^{\dagger} \tag{331}$$

For example

$$\left(\frac{i}{4}\frac{\partial}{\partial x}\right)^{\dagger} = \left(\frac{i}{4}\right)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger}$$
(332)

$$\left(\frac{i}{4}\frac{\partial}{\partial x}\right)^{\dagger} = \left(-\frac{i}{4}\right)\left(-\frac{\partial}{\partial x}\right) \tag{333}$$

$$\left(\frac{i}{4}\frac{\partial}{\partial x}\right)^{\dagger} = \frac{i}{4}\frac{\partial}{\partial x}$$
(334)

ii) Let *A* and *B* as two operators.

For example

$$(A+B)^{\dagger} = A^{\dagger} + B^{\dagger}$$
(335)
DALAL INSTITUTE(336)

$$\left(\inf \left(\frac{\partial}{\partial x} + \frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(\frac{\partial}{\partial x}\right)^{\dagger} + \left(\frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(\frac{\partial}{\partial x^2}\right)^{\dagger} + \left(\frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(\frac{\partial}{\partial x^2}\right)^{\dagger} =$$

$$\left(\frac{\partial}{\partial x} + \frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(-\frac{\partial}{\partial x}\right) + \left(\frac{\partial^2}{\partial x^2}\right)$$

$$\left(\frac{\partial}{\partial x} + \frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(-\frac{\partial}{\partial x} + \frac{\partial^2}{\partial x^2}\right)$$

$$(337)$$

$$(338)$$

iii) Let *A* and *B* as two operators, then

$$(AB)^{\dagger} = A^{\dagger}B^{\dagger} \tag{339}$$

For example

$$\left(\frac{\partial}{\partial x}\frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(\frac{\partial}{\partial x}\right)^{\dagger} \left(\frac{\partial^2}{\partial x^2}\right)^{\dagger}$$
(340)

$$\left(\frac{\partial}{\partial x}\frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(-\frac{\partial}{\partial x}\right)\left(\frac{\partial^2}{\partial x^2}\right)$$
(341)

D DALAL INSTITUTE

$$\left(\frac{\partial}{\partial x}\frac{\partial^2}{\partial x^2}\right)^{\dagger} = \left(-\frac{\partial^3}{\partial x^3}\right)$$
(342)

iv) Let A be the operators, then

$$\left(A^{\dagger}\right)^{\dagger} = A \tag{343}$$

For example

$$\left[\left(\frac{\partial}{\partial x}\right)^{\dagger}\right]^{\dagger} = \left(\frac{\partial}{\partial x}\right) \tag{344}$$

It should also be noted that the multiplication to an anti-hermitian operator by *i* makes it Hermitian, while the vice-versa is also equally true for adjoints.

v) For any operator A and its adjoint, the product (AA^{\dagger}) is Hermitian. For instance

$$\left(\frac{\partial}{\partial x}\right)\left(-\frac{\partial}{\partial x}\right) = -\frac{\partial^2}{\partial x^2}$$
(343)

vi) For any operator A and its adjoint, the sum $(A+A^{\dagger})$ is Hermitian. For instance

(343)
vii) For any operator A and its adjoint, then
$$AA^{\dagger} + A^{\dagger}A$$
 is Hermitian. For instance

$$(i3)(i3)^{\dagger} + (i3)^{\dagger}(i3) = (i3)(-i3) + (-i3)(i3) = 9 + 9 = 18$$
(343)

3. Characterization of Hermitian operator: We know that the average value of any operator (say \hat{A}) in quantum mechanics is calculated by the equation given below.

$$\bar{A} = \int \psi^* \hat{A} \psi dx \tag{344}$$

Where ψ is the wave function representing any quantum mechanical state and ψ^* is its complexes conjugate. Now because of the fact that the average value of any physical observable must be a real value, we can say that the operator used in equation (344) must follow the following condition.

$$\bar{A} = \bar{A}^* \tag{345}$$

$$\int \psi^* \hat{A} \psi dx = \left[\int \psi^* \hat{A} \psi dx \right]^*$$
(346)

$$\int \psi^* \hat{A} \psi dx = \int (\psi^*)^* (\hat{A} \psi)^* dx$$
(347)



$$\int \psi^* \hat{A} \psi dx = \int \psi (\hat{A} \psi)^* dx$$
(348)

Every linear operator that satisfies the equation (348) for all quantum-mechanically acceptable wave functions is called the Hermitian operator.

Besides the form given by equation (348), one more popular definition of a Hermitian operator is also given below.

$$\int f^* \hat{A} g dx = \int g (\hat{A} f)^* dx \tag{349}$$

From the equation, we can state that a Hermitian operator must fulfill the condition for the well-behaved functions f and g. It can be clearly seen that on the left side of the equation (349), \hat{A} is operated over the function g; while on the right side, the \hat{A} is operated over the function f. However, if we put f = g, the equation (349) is also reduced to equation (348); indicating that both definitions are correct.

4. Properties of Hermitian operators: The important properties of Hermitian operators are discussed below. *i) The eigenvalues of Hermitian operators are always real:*

Let \hat{A} be a Hermitian operator with a well-behaved wavefunction ψ representing a quantum mechanical state, then we can say that

Each side of equation (350) can be expressed as an imaginary and a real part as well; with left-hand real part equal to the right-hand real part, while left side imaginary part equal to right imaginary one. After taking the complex conjugate of equation (350), the imaginary parts would reverse sign but still holding the condition of equivalence.

$$\hat{A}^* \psi^* = a^* \psi^* \tag{351}$$

Multiplying the equation (350) by ψ^* and integrating over the whole configurational space, we get

$$\int \psi^* \hat{A} \psi dx = a \int \psi^* \psi dx \tag{352}$$

Similarly, multiplying the equation (351) by ψ and integrating over the whole configurational space, we get

$$\int \psi \hat{A}^* \psi^* dx = a^* \int \psi \psi^* dx \tag{353}$$

Now because left-hand sides of equation (352) and (353) are equal to each other (owing to the Hermitian nature of the operator), the right-hand sides are also equivalent; therefore, we can say that

$$a^* \int \psi \psi^* dx = a \int \psi^* \psi dx \tag{354}$$

$$0 = (a - a^*) \int \psi^* \psi dx \tag{355}$$

Since the wave function is a square-integrable, the integral part of the equation (355) cannot be zero and left us with the only possibility given blow.

$$(a - a^*) = 0 \tag{356}$$

$$a = a^* \tag{357}$$

The physical interpretation of the result given by equation (357) is that *a* must be real in order to yield zero from equation (356).

ii) Non-degenerate eigenfunctions of Hermitian operators are always orthogonal to each other:

Let ψ_m and ψ_n be two square-integrable eigenfunctions of a Hermitian operator \hat{A} ; therefore, we say

 $\hat{A}\psi_m = a_1\psi_m$

also

$$\hat{A}^*\psi_n^* = a_2\psi_n^*$$
(359)
Multiplying the equation (358) by ψ_n^* and integrating over the whole configurational space, we get
$$(1100000 \pm \int \psi_n^* \hat{A} \psi_m dx = a_1 \int \psi_n^* \psi_m dx$$
(360)

Similarly, multiplying the equation (359) by ψ_m and integrating over the whole configurational space, we get

$$\int \psi_m \hat{A}^* \psi_n^* dx = a_2 \int \psi_m \psi_n^* dx \tag{361}$$

Now because left-hand sides of equation (360) and (361) are equal to each other (owing to the Hermitian nature of the operator), the right-hand sides are also equivalent; therefore, we can say that

$$a_1 \int \psi_n^* \psi_m dx = a_2 \int \psi_m \psi_n^* dx \tag{362}$$

$$(a_1 - a_2) \int \psi_m \psi_n^* dx = 0$$
 (363)

Since the wave functions used are non-degenerate i.e. $a_1 \neq a_2$; the only possibility we are left with for the equation to be true is given below.

$$\int \psi_m \psi_n^* dx = 0 \tag{364}$$

Hence, we can say that ψ_m and ψ_n are definitely orthogonal to each other.



(358)

iii) If two Hermitian operators commute, their product is also a Hermitian operator:

Let ψ_1 and ψ_2 be two well-behaved functions; while \hat{A} and \hat{B} as two Hermitian operators. Therefore, we can say that

$$\int \psi_1^* \hat{A} \hat{B} \psi_2 dx \tag{365}$$

Since \hat{A} is Hermitian, we can say that

$$\int \psi_1^* \hat{A} \hat{B} \psi_2 dx = \int \psi_1^* \hat{A} (\hat{B} \psi_2) dx \tag{366}$$

$$\int \hat{A}^* \psi_1^* \hat{B} \psi_2 dx = \int \psi_1^* \hat{A} (\hat{B} \psi_2) dx$$
(367)

Since \hat{B} is also Hermitian, therefore

$$\int (\hat{A}^* \psi_1^*) \hat{B} \psi_2 dx = \int \hat{B}^* \hat{A}^* \psi_1^* \psi_2 dx$$
(368)

From equation (366) and (368), we get

$$\int \psi_1^* \hat{A} \hat{B} \psi_2 dx = \int \hat{B}^* \hat{A}^* \psi_1^* \psi_2 dx$$
(369)
(info@dalalinstitute.com, +91-9802825820)
commute with each other, we have

Therefore, equation (369) becomes

If the operator \hat{A} and \hat{B}

$$\int \psi_1^* \hat{A} \hat{B} \psi_2 dx = \int \hat{A}^* \hat{B}^* \psi_1^* \psi_2 dx \tag{371}$$

Which is the condition for the product operator to act as Hermitian.

iv) If two Hermitian operators do not commute, their commutator operator is anti-Hermitian in nature:

Let \hat{A} and \hat{B} as two Hermitian operators; therefore, we can say that their commutation must follow the following condition.

$$[\hat{A}, \hat{B}]^* = (\hat{A}\hat{B})^* - (\hat{B}\hat{A})^*$$
 (372)

or

$$\left[\hat{A},\hat{B}\right]^{*} = \hat{A}^{*}\hat{B}^{*} - \hat{B}^{*}\hat{A}^{*} = -\left(\hat{B}^{*}\hat{A}^{*} - \hat{A}^{*}\hat{B}^{*}\right)$$
(373)

or

Buy the complete book with TOC navigation, Copyright © Mandeep Dalal high resolution images and no watermark.



(370)

$$[\hat{A}, \hat{B}]^* = -[\hat{B}, \hat{A}]^*$$
(374)

$$\left[\hat{A}, \hat{B}\right]^{*} = -\left[\hat{B}, \hat{A}\right]^{*}$$
(375)

For instance, consider the commutator of position and momentum operator

$$[\hat{x}, \hat{p}_x] = i \frac{h}{2\pi} \tag{376}$$

The commutator $i\hbar$ is antihermitian in nature.

> The Linear Momentum Operator as Hermitian

In order to prove the linear momentum operator as the Hermitian, we must find its Hermitian conjugate first. The general expression of linear momentum operator is

$$\hat{p}_{x} = \frac{h}{2\pi i} \frac{\partial}{\partial x}$$
(377)
Let \hat{p}_{x}^{\dagger} be the Hermitian conjugate which can be calculated as follows:
$$\hat{p}_{x}^{\dagger} = \left(\frac{h}{2\pi i}\right)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger}$$
(378)
(378)
(378)
(379)
or
$$\hat{p}_{x}^{\dagger} = \left(-\frac{h}{2\pi i}\right) \left(\frac{1}{2\pi i}\right) \left(\frac{\partial}{\partial x}\right)$$
(379)
(380)

Comparing equation (377) and (380), we can see that the Hermitian conjugate of linear momentum operator is exactly equal to the linear momentum operator i.e. $\hat{p}_x^{\dagger} = \hat{p}_x$; proving that it is defiantly a Hermitian operator.

> The Angular Momentum Operator as Hermitian

In order to prove the angular momentum operator as Hermitian, we must find its Hermitian conjugate first. The general expression of the angular momentum operator is

$$\hat{L} = \frac{h}{2\pi i} \left[\left(y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right) + \left(z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right) + \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right) \right]$$
(381)

Let \hat{L}^{\dagger}_{χ} be the Hermitian conjugate which can be calculated as follows:





or

or

$$\hat{L}_{x}^{\dagger} = \left[\frac{h}{2\pi i} \left[\left(y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right) + \left(z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right) + \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right) \right] \right]^{\dagger}$$
(382)
$$= \left[\frac{h}{2\pi i} y \frac{\partial}{\partial z} - \frac{h}{2\pi i} z \frac{\partial}{\partial y} + \frac{h}{2\pi i} z \frac{\partial}{\partial x} - \frac{h}{2\pi i} x \frac{\partial}{\partial z} + \frac{h}{2\pi i} x \frac{\partial}{\partial y} - \frac{h}{2\pi i} y \frac{\partial}{\partial x} \right]^{\dagger}$$

or

$$\hat{L}_{x}^{\dagger} = \left(\frac{h}{2\pi i}\right)^{\dagger} (y)^{\dagger} \left(\frac{\partial}{\partial z}\right)^{\dagger} - \left(\frac{h}{2\pi i}\right)^{\dagger} (z)^{\dagger} \left(\frac{\partial}{\partial y}\right)^{\dagger} + \left(\frac{h}{2\pi i}\right)^{\dagger} (z)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger} - \left(\frac{h}{2\pi i}\right)^{\dagger} (x)^{\dagger} \left(\frac{\partial}{\partial z}\right)^{\dagger} + \left(\frac{h}{2\pi i}\right)^{\dagger} (x)^{\dagger} \left(\frac{\partial}{\partial y}\right)^{\dagger} - \left(\frac{h}{2\pi i}\right)^{\dagger} (y)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger}$$
(383)

or

$$\hat{L}_{x}^{\dagger} = \left(-\frac{h}{2\pi i}\right)(y)\left(-\frac{\partial}{\partial z}\right) - \left(-\frac{h}{2\pi i}\right)(z)\left(-\frac{\partial}{\partial y}\right) + \left(-\frac{h}{2\pi i}\right)(z)\left(-\frac{\partial}{\partial x}\right) \tag{384}$$

$$= \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial z}\right) + \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial y}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(y)\left(-\frac{\partial}{\partial x}\right) = \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial y}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(y)\left(-\frac{\partial}{\partial x}\right) = \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial y}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(y)\left(-\frac{\partial}{\partial x}\right) = \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial y}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(y)\left(-\frac{\partial}{\partial z}\right) = \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial y}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial z}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{\partial}{\partial y}\right)$$

$$= \left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{h}{2\pi i}\right)(x)\left(-\frac{h}{2\pi i}\right)($$

or

Comparing equation (381) and (386), we can see that the Hermitian conjugate of the angular momentum operator is exactly equal to the angular momentum operator i.e. $\hat{L}^{\dagger} = \hat{L}$; proving that it is defiantly a Hermitian operator.

> The Hamiltonian or Energy Operator as Hermitian

In order to prove the energy operator as Hermitian, we must find its Hermitian conjugate first. The general expression of the energy operator is

$$\widehat{H} = \frac{-h^2}{8\pi^2 m} \frac{\partial^2}{\partial x^2} + V(x)$$
⁽³⁸⁷⁾

Let \hat{H}^{\dagger} be the Hermitian conjugate which can be calculated as follows:



$$\widehat{H}^{\dagger} = \left[\frac{-h^2}{8\pi^2 m}\frac{\partial^2}{\partial x^2} + V(x)\right]^{\dagger}$$
(388)

or

$$\widehat{H}^{\dagger} = \left[\frac{-h^2}{8\pi^2 m}\frac{\partial}{\partial x}\frac{\partial}{\partial x} + V(x)\right]^{\dagger}$$
(389)

$$\widehat{H}^{\dagger} = \left(\frac{-h^2}{8\pi^2 m}\right)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger} \left(\frac{\partial}{\partial x}\right)^{\dagger} + (V(x))^{\dagger}$$
(390)

or

$$\widehat{H}^{\dagger} = \left(\frac{-h^2}{8\pi^2 m}\right) \left(-\frac{\partial}{\partial x}\right) \left(-\frac{\partial}{\partial x}\right) + (V(x))$$
⁽³⁹¹⁾

$$\widehat{H}^{\dagger} = \frac{-h^2}{8\pi^2 m} \frac{\partial^2}{\partial x^2} + V(x)$$
⁽³⁹²⁾

Comparing equation (387) and (392), we can see that the Hermitian conjugate of energy operator is exactly equal to the energy operator i.e. $\hat{H}^{\dagger} = \hat{H}$; proving that it is defiantly a Hermitian operator.



LEGAL NOTICE

This document is an excerpt from the book entitled "A Textbook of Physical Chemistry – Volume 1 by Mandeep Dalal", and is the intellectual property of the Author/Publisher. The content of this document is protected by international copyright law and is valid only for the personal preview of the user who has originally downloaded it from the publisher's website (www.dalalinstitute.com). Any act of copying (including plagiarizing its language) or sharing this document will result in severe civil and criminal prosecution to the maximum extent possible under law.



This is a low resolution version only for preview purpose. If you want to read the full book, please consider buying.

Buy the complete book with TOC navigation, high resolution images and no watermark.



Home: https://www.dalalinstitute.com/ Classes: https://www.dalalinstitute.com/classes/ Books: https://www.dalalinstitute.com/books/ Videos: https://www.dalalinstitute.com/videos/ Location: https://www.dalalinstitute.com/location/ Contact Us: https://www.dalalinstitute.com/contact-us/ About Us: https://www.dalalinstitute.com/about-us/

Postgraduate Level Classes		Undergraduate Level Classes		
(NET-JRF & IIT-GATE)		(M.Sc Entran	(M.Sc Entrance & IIT-JAM)	
Admission		Admission		
Regular Program	Distance Learning	Regular Program	Distance Learning	
Test Series	Result	Test Series	Result	

A Textbook of Physical Chemistry - Volume 1

"A Textbook of Physical Chemistry – Volume 1 by Mandeep Dalal" is now available globally; including India, America and most of the European continent. Please ask at your local bookshop or get it online here. READ MORE

Join the revolution by becoming a part of our community and get all of the member benefits like downloading any PDF document for your personal preview.

Sign Up



A TEXTBOOK OF PHYSICAL CHEMISTRY Volume I

MANDEEP DALAL



First Edition

DALAL INSTITUTE

Table of Contents

CHAP	TER 1	11
Qua	ntum Mechanics – I	11
*	Postulates of Quantum Mechanics	11
*	Derivation of Schrodinger Wave Equation	16
*	Max-Born Interpretation of Wave Functions	21
*	The Heisenberg's Uncertainty Principle	24
*	Quantum Mechanical Operators and Their Commutation Relations	29
*	Hermitian Operators – Elementary Ideas, Quantum Mechanical Operator for Linear Momentu Angular Momentum and Energy as Hermitian Operator	ım, 52
*	The Average Value of the Square of Hermitian Operators	62
*	Commuting Operators and Uncertainty Principle (<i>x</i> & <i>p</i> ; <i>E</i> & <i>t</i>)	63
*	Schrodinger Wave Equation for a Particle in One Dimensional Box	65
*	Evaluation of Average Position, Average Momentum and Determination of Uncertainty in Positi and Momentum and Hence Heisenberg's Uncertainty Principle	ion 70
*	Pictorial Representation of the Wave Equation of a Particle in One Dimensional Box and Influence on the Kinetic Energy of the Particle in Each Successive Quantum Level	Its 75
*	Lowest Energy of the Particle	80
*	Problems	82
*	Bibliography	83
CHAP'	TER 2	84
Ther	modynamics – I	84
*	Brief Resume of First and Second Law of Thermodynamics	84
*	Entropy Changes in Reversible and Irreversible Processes	87
*	Variation of Entropy with Temperature, Pressure and Volume	92
*	Entropy Concept as a Measure of Unavailable Energy and Criteria for the Spontaneity of Reacti	ion .94
*	Free Energy, Enthalpy Functions and Their Significance, Criteria for Spontaneity of a Process	98
*	Partial Molar Quantities (Free Energy, Volume, Heat Concept) 1	.04
*	Gibb's-Duhem Equation	08
*	Problems	11
*	Bibliography1	12

CHAP	CHAPTER 3	
Cher	nical Dynamics – I	113
*	Effect of Temperature on Reaction Rates	113
*	Rate Law for Opposing Reactions of Ist Order and IInd Order	119
*	Rate Law for Consecutive & Parallel Reactions of Ist Order Reactions	127
*	Collision Theory of Reaction Rates and Its Limitations	135
*	Steric Factor	141
*	Activated Complex Theory	143
*	Ionic Reactions: Single and Double Sphere Models	147
*	Influence of Solvent and Ionic Strength	152
*	The Comparison of Collision and Activated Complex Theory	157
*	Problems	158
*	Bibliography	159
CHAP'	ГЕК 4	160
Elect	rochemistry – I: Ion-Ion Interactions	160
*	The Debye-Huckel Theory of Ion-Ion Interactions	160
*	Potential and Excess Charge Density as a Function of Distance from the Central Ion	168
*	Debye-Huckel Reciprocal Length	173
*	Ionic Cloud and Its Contribution to the Total Potential	176
*	Debye-Huckel Limiting Law of Activity Coefficients and Its Limitations	178
*	Ion-Size Effect on Potential	185
*	Ion-Size Parameter and the Theoretical Mean - Activity Coefficient in the Case of Ionic C Finite-Sized Ions	louds with 187
*	Debye-Huckel-Onsager Treatment for Aqueous Solutions and Its Limitations	190
*	Debye-Huckel-Onsager Theory for Non-Aqueous Solutions	195
*	The Solvent Effect on the Mobility at Infinite Dilution	196
*	Equivalent Conductivity (Λ) vs Concentration $C^{1/2}$ as a Function of the Solvent	198
*	Effect of Ion Association Upon Conductivity (Debye-Huckel-Bjerrum Equation)	200
*	Problems	209
*	Bibliography	210
CHAP'	ΓER 5	211
Qua	ntum Mechanics – II	211
*	Schrodinger Wave Equation for a Particle in a Three Dimensional Box	211

*	The Concept of Degeneracy Among Energy Levels for a Particle in Three Dimensional Box	215
*	Schrodinger Wave Equation for a Linear Harmonic Oscillator & Its Solution by Polynomial N	Method 217
*	Zero Point Energy of a Particle Possessing Harmonic Motion and Its Consequence	229
*	Schrodinger Wave Equation for Three Dimensional Rigid Rotator	231
*	Energy of Rigid Rotator	241
*	Space Quantization	243
*	Schrodinger Wave Equation for Hydrogen Atom: Separation of Variable in Polar Sp	herical
	Coordinates and Its Solution	247
*	Principal, Azimuthal and Magnetic Quantum Numbers and the Magnitude of Their Values	268
*	Probability Distribution Function	276
*	Radial Distribution Function	278
*	Shape of Atomic Orbitals $(s, p \& d)$	281
*	Problems	287
*	Bibliography	288
CHAP	ГЕR 6	289
Ther	modynamics – II	289
*	Clausius-Clapeyron Equation	289
*	Law of Mass Action and Its Thermodynamic Derivation	293
*	Third Law of Thermodynamics (Nernst Heat Theorem, Determination of Absolute E	ntropy,
	Unattainability of Absolute Zero) And Its Limitation	296
*	Phase Diagram for Two Completely Miscible Components Systems	304
*	Eutectic Systems (Calculation of Eutectic Point)	311
*	Systems Forming Solid Compounds A _x B _y with Congruent and Incongruent Melting Points	321
*	Phase Diagram and Thermodynamic Treatment of Solid Solutions	332
*	Problems	342
*	Bibliography	343
CHAP	TER 7	344
Cher	nical Dynamics – II	344
*	Chain Reactions: Hydrogen-Bromine Reaction, Pyrolysis of Acetaldehyde, Decomposit	tion of 344
*	Photochemical Reactions (Hydrogen-Bromine & Hydrogen-Chlorine Reactions)	
*	General Treatment of Chain Reactions (Ortho-Para Hydrogen Conversion and Hydrogen-B	romine
•	Reactions)	358

*	Apparent Activation Energy of Chain Reactions	362
*	Chain Length	364
*	Rice-Herzfeld Mechanism of Organic Molecules Decomposition (Acetaldehyde)	366
*	Branching Chain Reactions and Explosions (H2-O2 Reaction)	368
*	Kinetics of (One Intermediate) Enzymatic Reaction: Michaelis-Menten Treatment	371
*	Evaluation of Michaelis's Constant for Enzyme-Substrate Binding by Lineweaver-Burk H Eadie-Hofstee Methods	Plot and 375
*	Competitive and Non-Competitive Inhibition	378
*	Problems	388
*	Bibliography	389
СНАР	TER 8	390
Elect	trochemistry – II: Ion Transport in Solutions	390
*	Ionic Movement Under the Influence of an Electric Field	390
*	Mobility of Ions	393
*	Ionic Drift Velocity and Its Relation with Current Density	394
*	Einstein Relation Between the Absolute Mobility and Diffusion Coefficient	398
*	The Stokes-Einstein Relation	401
*	The Nernst-Einstein Equation	403
*	Walden's Rule	404
*	The Rate-Process Approach to Ionic Migration	406
*	The Rate-Process Equation for Equivalent Conductivity	410
*	Total Driving Force for Ionic Transport: Nernst-Planck Flux Equation	412
*	Ionic Drift and Diffusion Potential	416
*	The Onsager Phenomenological Equations	418
*	The Basic Equation for the Diffusion	419
*	Planck-Henderson Equation for the Diffusion Potential	422
*	Problems	425
*	Bibliography	426
INDEX	ζ	427



Mandeep Dalal (M.Sc, Ph.D, CSIR UGC - NET JRF, IIT - GATE) Founder & Director, Dalal Institute Contact No: +91-9802825820 Homepage: www.mandeepdalal.com E-Mail: dr.mandeep.dalal@gmail.com Mandeep Dalal is an Indian research scholar who is primarily working in the field of Science and Philosophy. He received his Ph.D in Chemistry from Maharshi Dayanand University, Rohtak, in 2018. He is also the Founder and Director of "Dalal Institute", an India-based educational organization which is trying to revolutionize the mode of higher education in Chemistry across the globe. He has published more than 40 research papers in various international scientific journals, including mostly from Elsevier (USA), IOP (UK) and Springer (Netherlands).







Main Market, Sector-14, Rohtak, Haryana-124001 (+91-9802825820, info@dalalinstitute.com) www.dalalinstitute.com